Drugs for 3rd World Diseases: The Good, the Bad, and the Ugly

Richard Elliott

4th Annual CDD Community Meeting

Oct 21, 2010



The Good, the Bad, and the Ugly



The Ugly Tuco (Eli Wallach)

How do you define an ugly molecule?

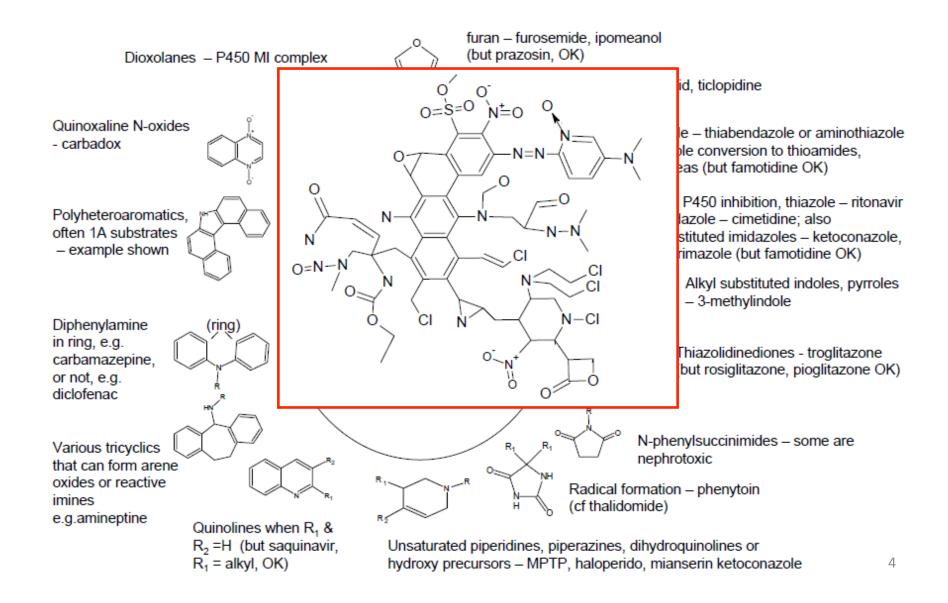
What do we really mean when we say a molecule is "ugly"? Can we really tell is a molecule is ugly just by looking at it?

The unsavory nature of 3rd world disease drugs

GRC Medicinal Chemistry Meeting, August 8-13, 2010; Colby-Sawyer College

7:30 pm - 9:30 pm	THIRD WORLD DISEASES
	Discussion Leader: Richard Elliott (Bill and Melinda Gates Foundation)
7:30 pm - 8:00 pm	Robert T. Jacobs (Scynexis) "Discovery and Pre-Clinical Development of SCYX-7158 (AN 5568), a Novel Oxaborole-6-Carboxamide with Potential for Treatment of Stage 2 Human African Trypanosomiasis"
8:00 pm - 8:10 pm	Discussion
8:10 pm - 8:40 pm	Jeremy Burrows (MMV) "Antimalarial Medicinal Chemistry: challenges & opportunities"
8:40 pm - 8:50 pm	Discussion
8:50 pm - 9:20 pm	Clifton Barry III (NIH) "Antitubercular nitroimidazoles: structure-based approaches to optimization of a prodrug"
9:20 pm - 9:30 pm	Discussion

Some Common Structural Alerts

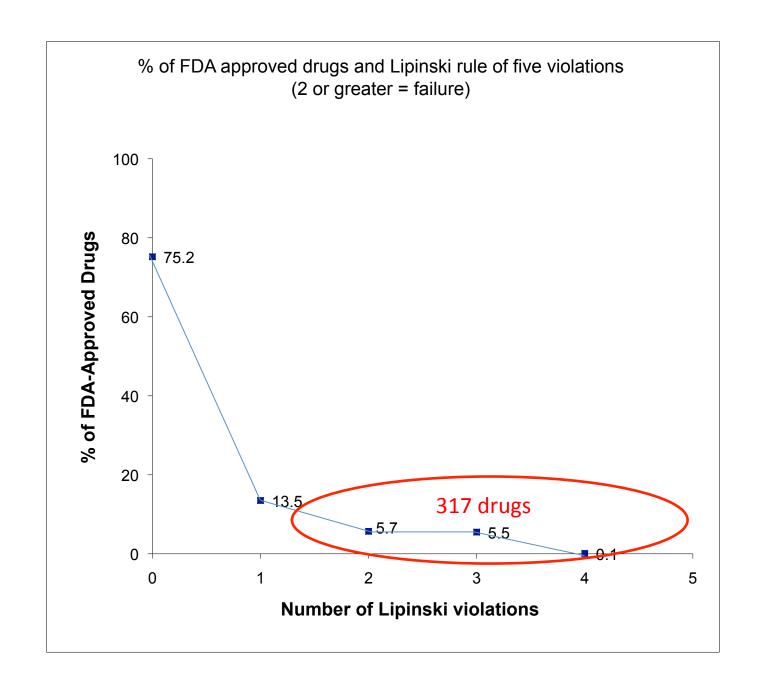


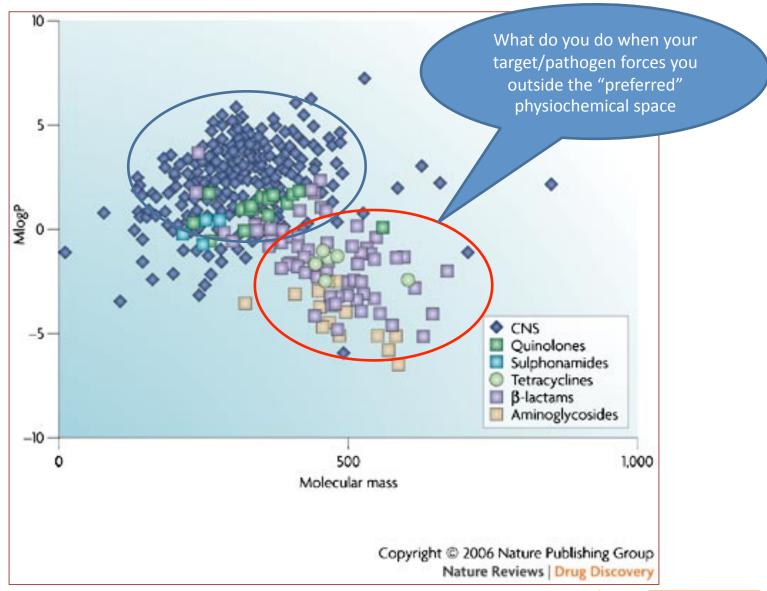
Lipinski's Rule-of-Five

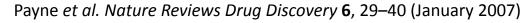
Lipinski's rule says that, in general, an orally active drug has no more than one violation of the following criteria:

- Not more than 5 hydrogen bond donors (nitrogen or oxygen atoms)
 with one or more hydrogen atoms)
- Not more than 10 hydrogen bond acceptors (nitrogen or oxygen atoms)
- A molecular weight under 500 daltons
- An octanol-water partition coefficient log P of less than 5

Are structural alerts/Lipinski rule's really the best to say if a molecule is ugly? How well do these rules work for third-world drug discovery

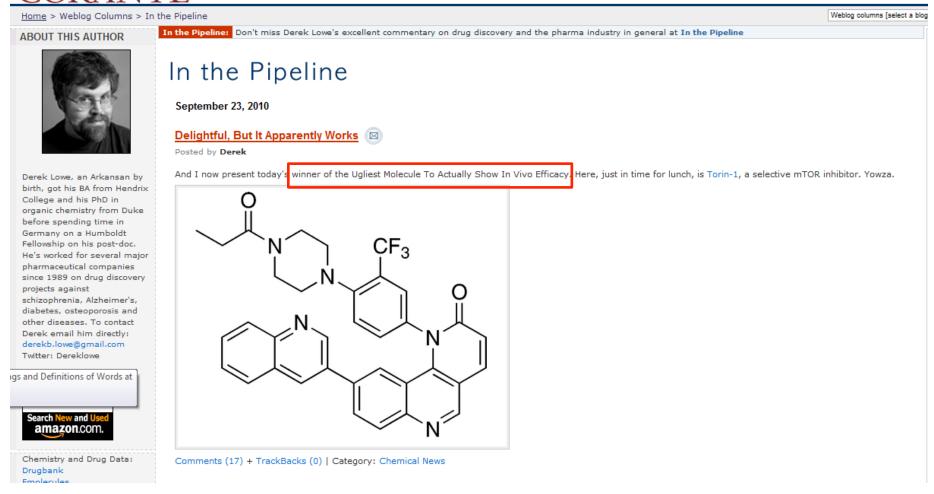








CORANTE



<u>Grayscale Rainbow on September 23, 2010 12:30 PM writes...</u> Oh, come on Derek, fess up. You know somewhere in your career you've made a compound or two (or more) with a MW~600 and cLogP ~6, which weren't even active in vitro, much less in vivo. Most of us have but we're too ashamed to admit it, unless of course one of them is in the clinic.

Mark on September 23, 2010 1:43 PM writes... My question is, what color is it?

Milkshake on September 23, 2010 12:35 PM writes... I think this one still has some solubility left because there are no free NH in the molecule. For another horrific example of an oral drug design, see itraconazole and posaconazole

Posaconazole

%F = >90% (food effect); $T_{1/2}$ = 35 hours

Posaconazole, initially available as an oral solution, was launched in European markets after gaining its first country approvals in December 2005; US FDA approval was received in September 2006 and analysts predict total sales of \$500 million in 2010.

Merck entering Phase II studies for Chagas Disease

http://www.merck.com/newsroom/news-release-archive/research-and-development/2010_0624.html

TB Drugs

Isoniazid

$$H_2N$$
 H_2
 H_2N
 H_2N
 H_3N
 H_4N
 H_2N
 H_2N
 H_3N
 H_4N
 H_2N
 H_2N
 H_3N
 H_4N
 H_5N
 H

Capreomycin

Rifampin

Anti-Malarial Drugs

From 13,500 GSK hits <1uM, average MWt 446 and clogP 5 (cf collection with average MWt 385 and clogP 3.3; [Gamo, F.-J. et al, Nature, **2010**, 465, 305]

Other 3rd World Drugs

But these are older drugs, what about the discovery pipeline?

What about our Discovery Pipeline of 3rd World Disease Drugs?

SCYNX-7158 (HAT)

OZ-277 (Malaria)

What about 1st world disease drugs?



http://www.drugs.com/top200.html

Some Basic Questions

- 1. Why do drugs for 3rd world diseases look like this?
- 2. If these molecules have higher risk, how do we manage these risks?
- 3. Are there any lessons from 3rd world disease drugs?
- 4. Have we become too dogmatic in how we judge molecules?

1) Why do drugs for 3rd world diseases look like this?

- Age of drugs
- Historical (natural) source of drugs
- Nature of the beast:
 - Permeability barriers (waxy cell walls, intracellular location)
- Activatable "warheads" for maximal activity
 - Nitro groups, endoperoxides, acylhydrazides
 - Hit multiple targets; minimize resistance, kill quickly
 - Use parasitic machinery for selective activation
- Non specific mechanisms
 - Membrane disruption
 - Cellular accumulation

Proposed Mechanism of Action of Some Commonly Used Third-World Disease Drugs

inhibits protein synthesis by binding to the 30S ribosomal unit
forming a transmembrane channel that leads to monovalent ion leakage
endoperoxide activated to reactive intermediate(s)
mitochondrial electron transport, ATP and pyrimidine biosynthesis.
inhibits protein synthesis by binding to the 70S ribosomal unit
binds to the guanine bases of bacterial DNA
activated by catalase-peroxidase enzyme (KatG) to reactive intermediate
binds and activates glutamate-gated chloride channels
not well defined; perhaps inhibits the formation of ß-hematin
disruption of energy generation due to the high affinity for sulfhydryl groups
bioreductive activation of nitro group
Bio-reductive activation of nitro group
inhibits dihydrofolate reductase (prodrug), perhaps other targets
inhibit mycobacterial transcription by targeting DNA-dep. RNA polymerase

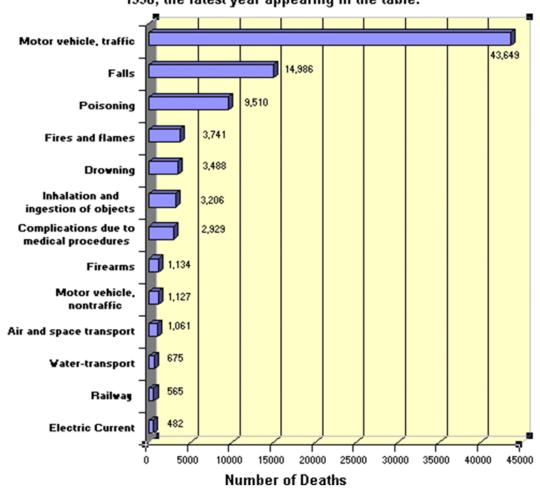
2) If these molecules have higher risk, how do we manage these risks?

- Utilize modern state-of-the-art risk-management tools and approaches
 - Computational
 - In vitro
 - In vivo
- Develop new risk-management tools and methods
- Adapt a portfolio approach to the discovery pipeline
- Make data-driven decisions regarding series, compounds

Without Data, Risk Assessment can be Subjective

Deaths from Accidents, by Type

Source: U. S. Census Bureau, Statistical Abstract of the United States: 1999, Page 106, table 146. Figures are for 1996, the latest year appearing in the table.





"Yes, I used to trade stocks, but I found it too risky."

3) Are there any lessons from 3rd world disease drugs?

- Don't be too proscriptive/dogmatic in assessing risks- keep an open mind.
- We have explored much less chemical diversity than we think we have.
- We need new chemistry to access addition chemical diversity.
- Our risk assessment tools are still inadequate
- We still have much to learn from nature!

4) Have we become too dogmatic in how we judge molecules?

Yes (we tend to be):

- We need to remain open-minded about the physiochemical space and chemotypes we are willing to explore
- We have developed "rules" rather than "guidelines"
- We have under-estimated the risks of not taking risks
- We still aren't really able to distinguish good from bad/ugly.

Question: How do you define an ugly molecule?

Answer: An ugly molecule should be defined by what it does, not by how it looks

The End

Drugs for 3rd World Diseases: The Good, the Bad, and the Ugly

Over the last 15 years numerous empirically-derived rules, guidelines, and tools based on retrospective analyses have been developed to enable medicinal chemists to better understand and more effectively manage drug development risks associated with a drug candidate's chemical structure and physiochemical properties. These analyses have led to a greater consensus within the drug development community regarding acceptable physiochemical properties and structural motifs for drugs. However, analysis of current drugs used for the treatment of 3rd world diseases, most of which are safe and efficacious, and many of which have been dosed in millions of patients, reveals a multitude of "forbidden" functionalities (structural alerts) and "unacceptable" properties, including endoperoxides, nitro-aromatics, and large macrocyclic ring systems which fall well outside the canonical "rule of five" boundaries. This raises a number of fundamental questions, including does drug discovery for 3rd world diseases require different thinking and risk-tolerance regarding molecular properties and chemical functionalities from 1st world disease drug discovery? If so, how can we further understand these differences and capitalize on this knowledge to better develop and risk-manage the 3rd world drug discovery portfolio?